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Xiao-Dan Cao,^a Dong-Zhi Liu,^a* Xue-Qin Zhou^b and Shao-Feng Sun^a

^aSchool of Chemical Engineering and Technology, Tianjin University, Tianjin 300072, People's Republic of China, and ^bSchool of Materials Science and Engineering, Tianjin University, Tianjin 300072, People's Republic of China

Correspondence e-mail: cherrycao@eyou.com

Key indicators

Single-crystal X-ray study T = 293 K Mean σ (C–C) = 0.010 Å R factor = 0.056 wR factor = 0.154 Data-to-parameter ratio = 7.1

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

{4-[(1-Naphthyl)vinyl]phenyl}di-p-tolylamine

The title compound, $C_{32}H_{27}N$, was synthesized *via* the Ullmann reaction. The structure has a pseudo-twofold axis approximately parallel to the *b* axis, and hence the pseudo-space group C2/c. The crystal packing is stabilized by weak $C-H\cdots\pi$ and and van der Waals interactions.

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Comment

Organic photoconductive materials are a new type of high technology information material which can generate electronhole pairs upon illumination. They have been extensively used in copier applications, laser printing and digital xerography (Yang & Geize, 1992). The title compound, (I), a triarylamine, is a charge transfer material which can be used in doublelayered photoconductive devices. We describe its structure here.



The molecular structure of (I) is illustrated in Fig. 1. The structure has a pseudo-twofold axis passing through the atoms N1, C15, C18, C21, C23 and C26, approximately parallel to the *b* axis, and hence it has the pseudo-space group C2/c. The two



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Figure 2

The crystal structure of (I), viewed along the c axis

toluene rings (C1–C6 and C8–C13) are oriented at angles 62.7 (3) and 68.0 (3)°, respectively, to the central aromatic ring (C15–C20). The dihedral angle between the naphthalene ring system [planar within 0.035 (1) Å] and the C15–C20 ring is 33.7 (2)°. The molecular packing in the crystal is stabilized by weak C–H··· π interactions (Table 1) and van der Waals interactions.

Experimental

4-(1-Naphthylvinyl)phenylamine (0.01 mol), 4-methyliodobenzene (0.026 mol), CuCl (0.002 mol), 1,10-phenanthroline (0.001 mol) and KOH (24 g) were dissolved in toluene (30 ml). The mixture was refluxed for 6 h and, after evaporation of the solvent, the residue was separated by column chromatography (silica gel, ethyl acetate/ petroleum ether = 1:200) to give the title compound (Pautmeier *et al.*, 1990). The compound was identified by IR and MS. Single crystals were obtained by slow evaporation of a petroleum ether solution; m.p: 383–386 K; MS: 425.3, 286.2, 244.1, 91.7, 51.6.

Crystal data

C32H27N
$M_r = 425.55$
Monoclinic, Cc
a = 18.482 (7) Å
b = 17.045 (7) Å
c = 8.635 (3) Å
$\beta = 117.094 \ (5)^{\circ}$
$V = 2421.9 (16) \text{ Å}^3$
Z = 4

 $D_x = 1.167 \text{ Mg m}^{-3}$ Mo K\alpha radiation Cell parameters from 724 reflections $\theta = 2.4-21.7^{\circ}$ $\mu = 0.07 \text{ mm}^{-1}$ T = 293 (2) K Block, yellow $0.40 \times 0.34 \times 0.28 \text{ mm}$

Data collection

Bruker SMART CCD area-detector diffractometer ω scans	2130 independent reflections 1453 reflections with $I > 2\sigma(I)$ $R_{int} = 0.028$
Absorption correction: multi-scan (SADABS; Bruker, 1997) $T_{min} = 0.973, T_{max} = 0.981$ 5052 measured raflections	$\theta_{\text{max}} = 25.0^{\circ}$ $h = -21 \rightarrow 21$ $k = -20 \rightarrow 10$ $h = -10 \rightarrow 7$
Refinement	
Refinement on F^2 $R[F^2 > 2\sigma(F^2)] = 0.056$ $wR(F^2) = 0.154$ S = 1.06 2130 reflections 300 parameters H-atom parameters constrained	$w = 1/[\sigma^{2}(F_{o}^{2}) + (0.0877P)^{2} + 0.2832P]$ where $P = (F_{o}^{2} + 2F_{c}^{2})/3$ $(\Delta/\sigma)_{max} = 0.001$ $\Delta\rho_{max} = 0.18 \text{ e} \text{ Å}^{-3}$ $\Delta\rho_{min} = -0.17 \text{ e} \text{ Å}^{-3}$

Table 1

Hydrogen-bonding geometry (Å, °).

$D - H \cdot \cdot \cdot A$	$D-\mathrm{H}$	$H \cdot \cdot \cdot A$	$D \cdots A$	$D - \mathbf{H} \cdots A$
$C2-H2\cdots Cg3^{i}$	0.93	3.09	3.835 (7)	138
$C10-H10\cdots Cg1^{ii}$	0.93	3.26	4.107 (8)	153
$C16-H16\cdots Cg2^{i}$	0.93	3.15	3.586 (6)	111
$C20-H20\cdots Cg2^{iii}$	0.93	3.03	3.592 (6)	121
$C30-H30\cdots Cg1^{iv}$	0.93	3.30	4.079 (11)	142

Symmetry codes: (i) $x, 1 - y, \frac{1}{2} + z$; (ii) $x, -y, \frac{1}{2} + z$; (iii) $x, 1 - y, z - \frac{1}{2}$; (iv) $\frac{1}{2} + x, \frac{1}{2} + y, z$. Notes: *Cg*1, *Cg*2 and *Cg*3 denote the centroids of the C8–C13, C23–C28 and C27–C32 rings, respectively.

H atoms were positioned geometrically, with C—H = 0.93–0.96 Å, and refined using a riding model, with $U_{iso}(H) = 1.2U_{eq}(C)$. Restraints for similarity and approximate isotropy were applied to the U_{ij} components of some of the naphthalene ring atoms. The Friedel pairs were merged during the final cycles of refinement.

Data collection: *SMART* (Bruker, 1997); cell refinement: *SMART*; data reduction: *SAINT* (Bruker, 1997); program(s) used to solve structure: *SHELXS*97 (Sheldrick, 1997); program(s) used to refine structure: *SHELXL*97 (Sheldrick, 1997); molecular graphics: *SHELXTL* (Bruker, 1997); software used to prepare material for publication: *SHELXTL*.

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